

# डॉ बी आर अम्बेडकर राष्ट्रीय प्रौद्योगिकी संस्थान,जालंधर

# Dr. B R Ambedkar National Institute of Technology, Jalandhar Department of Chemical Engineering

Ref.: NITJ/CH/ 912-

Date: 06 | 9 | 2022

Subject: Minutes of 2<sup>nd</sup> Internal Meeting of Board of Studies (BOS)

Dear Sir

Please find the Minutes of 2<sup>nd</sup> Internal Meeting of Board of Studies, Department of Chemical Engineering.

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MINUTES

OF

2nd INTERNAL MEETING OF

BOARD OF STUDIES

OF

DEPARTMENT OF CHEMICAL ENGINEERING



August 03, 2022 AT 12:30 PM DEPARTMENTAL LIBRARY

Dr. B.R.AMBEDKAR NATIONAL INSTITUE OF TECHNOLOGY JALANDHAR- 144011

2<sup>nd</sup> INTERNAL BOARD OF STUDIES (BOS) MEETING OF DEPARTMENTAL FACULTY MEMBERS WAS CONDUCTED ON AUGUST 03, 2022 AT 12:30 PM IN THE DEPARTMENTAL LIBRARY OF CHEMICAL ENGINEERING DR. B.R. AMBEDKAR NAIONAL INSTITUTE OF TECHNOLOGY, JALANDHAR. MINUTES OF THE MEETING OF BOS ARE AS FOLLOWS

### The following members were present:

Dr. Renu Gupta
 Associate Professor & Head
 Department of Chemical Engineering

Chairperson

 Dr. B.S. Kaith Professor Department of Chemistry Member

Dr. M.K. Jha
 Professor (HAG)
 Department of Chemical Engineering

Member

Dr. Sangeeta Garg
 Associate Professor
 Department of Chemical Engineering

Member

Dr. Poonam Gera
 Associate Professor
 Department of Chemical Engineering

Member

Dr. S. Bajpai
 Associate Professor
 Department of Chemical Engineering

Member

Associate Professor Department of Chemical Engineering

7. Dr N K Srivastava

Member

Dr. Jatinder Kumar Ratan
 Associate Professor
 Department of Chemical Engineering

Member

Dr. Neetu Divya
 Assistant Professor
 Department of Chemical Engineering

Member

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Member 10. Dr. Shashikant Yadav Assistant Professor Department of Chemical Engineering Member 11. Dr. Deepak Sahu Assistant Professor Department of Chemical Engineering Member 12. Dr. Amit D Saran Assistant Professor Department of Chemical Engineering Member 13. Dr. Anurag Tiwari Assistant Professor Department of Chemical Engineering Member 14. Dr. D Giri Babu Assistant Professor Department of Chemical Engineering Member 15. Dr. Nitin Naresh Pandhare Assistant Professor Department of Chemical Engineering Member Secretary 16. Dr. Anjireddy Bhavanam Assistant Professor Department of Chemical Engineering

The following members could not attend the meeting due to pre-occupation.

1. Dr Ajay Bansal

Member

Professor and Registrar (I/C)
Department of Chemical Engineering

Dr Harsh K Verma
 Professor and Dean Academic
 Computer Science & Engg. Department

Member

Dr. Renu Dhir
 Associate Professor
 Computer Science & Engg. Department

Member

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The Chairman, Board of Studies welcomed the members of the BOS. Thereafter, the following agenda items were taken up for discussion

Item No.	Particular No.
Item 2.1	Confirmation of Minutes of 1st Internal BOS meeting held on 19th May 2022
Item 2.2	Revision of Eligibility Criteria for Admission in M. Tech Chemical Engineering
Item 2.3	Revision of Eligibility Criteria for Admission in PhD Chemical Engineering
Item 2.4	Revision of Program Outcomes (POs) and Program Specific Outcomes (PSOs) of B.Tech. Chemical Engineering
Item 2.5	Revision of Course Outcomes (COs) Mapping with POs and PSOs of B.Tech. Chemical Engineering

### Item 2.1

Confirmation of Minutes of 1st Internal BOS meeting held on 19th May 2022.

Members approved the Minutes of 1st Internal BOS meeting held on 19th May 2022.

### Item 2.2

Revision of Eligibility Criteria for Admission in M. Tech Chemical Engineering Members deliberated and approved the revision of Eligibility Criteria for Admission in M. Tech Chemical Engineering. The finalized eligibility criteria is as follows

# Admission through CCMT\*

B.Tech in Engineering /Technology with valid GATE score

or

M.Sc in sciences with valid JRF-NET /GATE

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B.Pharm with valid GATE /G-PAT Score

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# Admission through Self-sponsored\*

B.Tech in Engineering /Technology

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MSc in sciences

or

B.Pharm

\*Other terms and conditions are as per the institute norms

### Item 2.3

# Revision of Eligibility Criteria for Admission in PhD Chemical Engineering

Members deliberated and approved the revision of Eligibility Criteria for Admission in PhD Chemical Engineering. The finalized eligibility criteria is as follows

Institute fellowship\*

M. Tech in Engineering/Technology along with Bachelor's Degree in Engineering/ Technology

B.Tech in Engineering /Technology with CGPA of 8.5 and above on a 10 point scale or 75% aggregate from a Centrally Funded Technical Institute (CFTI).

or

M.Sc in sciences with JRF-NET/GATE Score

M.Pharm with GATE /G-PAT Score

### Part time\*

M.Tech in Engineering/Technology along with Bachelor's Degree in Engineering/ Technology

Or

B. Tech in Engineering / Technology with CGPA of 8.5 and above on a 10 point scale or 75% aggregate from a Centrally Funded Technical Institute (CFTI).

MTech in Engineering / Technology

Or

MSc in sciences

Or

M.Pharm

Self sponsored\*

M. Tech in Engineering / Technology along with Bachelor's Degree in Engineering/ Technology

or

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B.Tech in Engineering /Technology with CGPA of 8.5 and above on a 10 point scale or 75% aggregate from a Centrally Funded Technical Institute (CFTI).

or

MSc in sciences

Or

M.Pharm

\*Other terms and conditions are as per the institute norms

### Item 2.4

# Revision of Program Outcomes (POs) and Program Specific Outcomes (PSOs) of B.Tech. Chemical Engineering

Members approved the revised Program Outcomes (POs) and Program Specific Outcomes (PSOs) of B.Tech. Chemical Engineering. The Revised POs and PSOs are as follows

## Programme Outcomes (POs):

Graduates of Chemical Engineering will demonstrate:

- 1. Engineering knowledge: Students will be able to learn the basics of various unit operations and processes related to chemical and allied industries.
- 2. **Problem analysis**: Students to identify formulate and analyse simple and complex Chemical Engineering problems.
- 3. Design/development of solutions: Students will be able to design solutions for complex engineering problems and design system components or processes that meet the specified needs with appropriate consideration for the public health and safety, and the cultural, societal, and environmental considerations.
- 4. Conduct investigations of complex problems: Use research-based knowledge and research methods including design of experiments, analysis and interpretation of data, and synthesis of the information to provide valid conclusions.
- Modern tool usage: Students will be able to learn and apply latest engineering softwares and skills for modelling, simulation and design of chemical engineering processes.
- 6. The engineer and society: Students to understand ethical aspects of chemical engineering and the impact of the profession on society.
- 7. Environment and sustainability: Students will have an understanding of global economic, environmental, demographic and political issues, the impact of engineering decisions on the local and global environment, economy, and society and cultures other than that from which they originate.

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- 8. Ethics: Students will be able to apply ethical principles and commit to professional ethics and responsibilities and norms of the engineering practice.
- Individual and team work: Students will be required to do the various curricular and extra-curricular activities individually and in groups to have multifaceted growth.
- 10. Communication: Students will be able to produce effective written and oral communication and adapt their presentation style and content to match the audience.
- 11. Project management and finance: Students will be able to apply knowledge and understanding of the engineering and management principles and implement these to their own work, as a member and leader in a team, to investigate the technoeconomic feasibility and to manage projects.
- 12. Life-long learning: Students will be proficient in the use of a variety of informational and educational media such as textbooks, scientific and technical journals, library system, internet and educational software, have an understanding of and exposure to the breadth and structure of the professional and technical support system that will be available to the students upon graduation.

# Programme Specific Outcomes (PSOs):

:

- PSO1 Apply the fundamentals of mathematics, basic science and engineering knowledge to understand, analyze, and solve the complex problems of chemical Engineering and allied disciplines.
- 2. **PSO2** To design, innovate and implement the knowledge for the benefit of chemical and allied industries.
- 3. **PSO3** Implementation of communication skills, professional engineering solutions and ethics for sustainable development of the society.

# Item 2.5 Revision of Course Outcomes (COs) Mapping with POs and PSOs of B.Tech. Chemical Engineering

Members approved the revised Course Outcomes (COs) Mapping with POs and PSOs of B.Tech. Chemical Engineering. The revised mapping is given Annexure I.

Chairman BOS may kindly go through it and approve the same.

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(Dr. Anjireddy Bhavanam)

Assistant Professor and Member Secretary

BOS, Chemical Engineering.

HD/CH Pally 109/22

Cum Chairperson, BOS, Chemical Engineering.

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# CO and PO mapping - Annexure I

34			CHPC-203			ų.	CHPC-201	Course Code
			Fluid Mechanics				Chemical Process Calculations	Course Title
CO4: Applying the principles of fluid mechanics to	CO3: Ability to analyze the fluid flow problems with the application to the momentum balance.	CO2: Ability to select pumps, valves, and would be able to calculate power requirement for pumping as well as agitation operations.	CO1: students would be able to measure pressure drop, flow rates etc. for incompressible and compressible fluids.	CO4: Understand the ideal and real behavior of gases, vapors and liquids	CO3: Able to draw the flow diagram and solve the problems involving recycle, purge and bypass in a process or unit	CO2: Able to perform material and energy balances on chemical processes/equipment without and with reactions.	erstand the nd energy of chemical	Course Outcomes
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	cO4:To understand the working and applications of	ilquias	agitation and mixing of	processes involved in	CO3: To understand the	solid-fluid mixtures.	processes for solid-solid and	understand separation	CO2: The student would	mixtures.	of solid-solid and solid-livin	measurement and handling	properties, property	understand the physical	CO1: The student would	problems.	construction and corrosion	knowledge on materials of	CO4: Ability to get	problems.	associated engineering	to understand the various	industries and	different chemical process	process flow diagrams of	CO3: Able to identify	processes involved.	unit operations and unit	CO2: Able to understand the	process.	qualitatively in each step of	quantitatively and	finished products	flow of raw materials to	CO1: Able to analyze the	problems.	chemical engineering
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	Fluid Mechanics Lab				Material Science for Chemical Engineers	
CO2 Students will be able to understand the fluid dynamics and also the principles of turbulent flow, calculation of turbulent boundary layers with pressure gradient, transition from	CO1 The student must be able to approach and solve typical problems in fluid dynamics at the appropriate level.	cO4: Able to get knowledge on advanced materials and their application.	in the metallic and nonmetallic material selection and handling material in chemical engineering in the areas of	engineering materials CO2: Understand the various properties and corrosion behavior of the selected materials in chemical industries.	CO1: Understand the basics knowledge such as internal structure, crystal geometry, crystal	solid-storage and conveying, and flow through packed and fluidized beds
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8	Chemical Engineering Thermodynamics			Operations Lab		
temperature, pressure, system, properties, process, state, cycles and equilibrium. CO2:Ability to apply the laws of thermodynamics for solving problems related to flow. processes and equilibrium	fluidized beds CO1:Ability to understand the basic concepts of thermodynamic such as	CO4: To experimentally analyze the working and applications of packed and	CO3: To experimentally analyze the parameters involved in agitation and mixing of liquids	analyze the properties, size- reduction, filtration and handling of solid-solid and solid-fluid mixtures CO2: To experimentally analyze separation processes for solid-solid and solid-fluid mixtures	CO4 Ability to understand and analyze the applications to industrial flows.	tomeasurevolumetric and mass flow rates through the Venturi meter and Orifice meter and efficiency of pumps.
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Chemical Technology-II				Heat Transfer		
CO1:To understand the manufacturing processes in various chemical process industries. etc. for incompressible and compressible fluids.	CO4: Ability to design and analyze the performance of heat exchangers and evaporators, reactor heating and cooling systems.	CO3: Develop correlations using elementary dimensional analysis and comprehend the laws governing radiation mode.	cO2: students should be able to perform heat flux calculations through constant and variable area elements and estimate optimum insulation thickness	CO1: Ability to understand and solve conduction, convection and radiation	CO4:Ability to understand the homogeneous gas phase reactions and heterogeneous chemical equilibrium	cO3:Ability to define energy transfer through mass, heat and work for closed and control volume systems.
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CO1: Introduce to standards	importance in fulfilling the present day energy needs.	technologies and their	alternate energy	knowledge on various	CO4 : Able to acquire the	utilization.	gaseous fuels and their	manufacturing process of	knowledge on	CO3 : Able to gain	petroleum refining.	process involved in	different types of unit	CO2 : Ability to understand	energy production.	conversion technologies for	preparation and their	coal origin, classification,	understand the concepts of	CO1 : The students will	problems.	construction and corrosion	knowledge on materials of	CO4: Ability to get	problems	associated engineering	to understand the various	industries and	different chemical process	process flow diagrams of	CO3: Able to identify	variations	industrial	process parameters and	different raw materials,	CO2: To understand the
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equipment at laboratory	exhibit the skills of handling	CO1:The students will	of a metals.	samples and corrosion rate	adulteration of food	nature of sugar sample,	reducing and non reducing	CO4: Ability to estimate the	preparation of urea.	analysis of cement and	CO3: Acquire knowledge in	viscometer	Molecular weight using	properties like viscosity,	measurement of various	CO2: Able to understand the	oil sample	saponification value of given	to determine acid and	CO1: Acquire the knowledge	from this course.	on the knowledge obtained	industrial equipment based	identify design problems in	design, fabricate and	CO4: Students will able to	testing as a designer.	techniques and equipment	understanding of fabrication	demonstrate general	CO3: The students would	design codes.	the students to use the	CO2: This course enables	process industry.	equipment used in the	for the mechanical design of
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CO4: Applying the concepts to mass transport to industrial flows.	calculations of number of transfer units and height of a transfer unit (NTU & HTU) concepts, packed column for absorption, equipment for gas absorption and batch & continuous drying equipment's	CO2: The students are able to comprehend the concepts of co current & counter current processes, cascades and concept of Ideal stage and stage efficiencies, continuous contact equipment's	CO1: Ability to understand the principles of mass transport	CO4: The students will develop the skill of presenting the results in form of written reports.	CO3: Ability to understand, explain and select instrumental for heat transfer analysis.	CO2: The students will develop the hand experience for industry.	scale and co-relate the theoretical aspects by performing experiments related to heat transfer.
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			Environmental Engineering Lab			
CO4: The students will develop the skill of	cO3: Students will be able to distinguish between the poor and good quality of life sustaining elements (water, air and soil).	CO2: Students can able to determining various properties contributing towards the water quality such as acidity, alkalinity, hardness, dissolved oxygen (DO), biochemical oxygen demand (BOD) and chemical oxygen demand(COD) present in the given sample	CO1: Students will develop the skills to co-relate the theoretical aspects by performing experiments.	CO4: The students exhibit the skill to solve the problems related to the environmental engineering.	CO3: The students gain the knowledge of different standards for the measure and control of air, water and solid waste pollution in the environment.	CO2: The students are able to design various engineering systems of control of air, water and solid waste pollution by equipment and by process changes.
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Reaction	Chemical		-																	Mass Transfer- II		<del>- 11 - 11</del>									•				l echilology Lab	Tachpology lah	Energy	. *:	
analyze the non- ideal flow	CO1To understand and	equipments	NTU for the adsorption	calculations of HETP and	CO4: Ability to perform	-liquid extraction operation	number of stages for liquid	recovery of solute and	able to calculate percentage	CO3 : The students will be	distillation country	of number of stages in a	metrion for determination	mothed for determination	able to apply McCabe Thiele	CO2: The students will be	types of distillation	distillation, methods and	the basic principles of	CO1:Ability to understand	residue	gases and determine carbon	CO4:Able to analyze the flue	process	application of distillation	knowledge on the	CO3: Able to gain	analysis for solid fuels	importance of proximate	CO2:Able to understand the	point	aniline, cloud and pour	, tire ,smoke, melting,	properties of fuels like flash	to illeasule valious	to measure various	CO1:The students will able	form of written reports.	presenting the results in
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							Lab	Mass Transfer								***************************************		5	***************************************					Control	Dynamics and	Process					-						×	Engineering-II
	related to mass	performing experiments	theoretical aspects by	CO2:Ability to co-relate the	laboratory scale.	the equipment's at	develop the skills to handle	CO1:The students will	and allied industries.	control system for chemical	to design the	control systems and be able	identified the stability of	CO4:The students can	controllers	response of various	to understand the transient	CO3:The students will able	higher order systems	first and	various forcing function on	to understand the effect of	CO2:The students will able	block diagram	chemical process in terms of	CO1: To understand the	reactor design	Fluid-Fluid reactions and	analyze kinetic regimes in	CO4To understand and	of solid catalysts	properties and preparation	CO3To understand physical	catalytic reactionsystems	internal transport in	analyze the external and	CO2To understand and	behavior in reactors
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		CHCI-300				CHPC-324				
		Minor Project, Phase-II			Lab	Reaction Engineering and Thermodynamics				
communicate effectively.	CO2: Ability to follow professional ethics while preparing project reports.	col: Able to use the chemical engineering knowledge to come up with innovative research proposals for growth of profession and society.	CO4: To experimentally verify the adsorption isotherms	CO3: To analyze the residence time distribution curve in a packed-bed reactor	CO2: Ability to analyze the properties of solid catalyst particles	CO1: Ability to understand and analyze the rate kinetics for the given reaction	CO4:Ability to understand and analyze the applications to industrial flows	column, agitated batch crystallize, liquid-liquid extraction and heat and mass balance in cooling tower can lead to design of the experiments.	CO3:The knowledge related to distillation column, liquid-liquid extraction in a packed	transfer operations.
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	based equipment's like heat	to handle the design of	CO1: The students are able	the real plant unit	accident and loss statics for	CO4:Able to calculate the	systems	prevention and control	CO3:Able to design the fire	Index for the real plant unit	Dow's fire and Explosion	such as conducting	performing risk assessment	exhibit the skill of	CO2:The students learn to	toxic dispersion etc.	such as pool fire, fireball,	hazard models	of loss prevention and	to understand the concept	CO1:The students are able	limitations.	solutions along with their	steady and time dependent	CO4: Ability to develop	conditions.	with appropriate boundary	industrial problems along	CO3: Ability to analyze	transfer analysis.	mass and momentum	CO2: Ability to do heat,	transport processes.	CO1: Understanding of	issues.	environmental and societal	CO4: Ability to address the
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Chemical Engineering Computing Lab		Process Control	
CO1:The students would be well versed with the principles of computing methods with the theory involved in the solving the chemical engineering problems.  CO2:The students would be able to independently solve the problems in the	CO2:Able to co-relate the theoretical aspects by performing experiments of process control i.e. liquid level tank and control valve characteristics.  CO3:Experimental exposure to the first and second order process systems.	CO4: The students will develop the skill of presenting the data in correct form CO1:The students will develop the skills to handle the equipments at laboratory scale	columns etc CO2: The students would also able to make plant layout of the newly developed plants and prepare written reports of design problems CO3: Ability to understand, explain chemical process plant.
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simulation with mathematical techniques	concepts of process model formulation, analysis of variables, parameter estimation and	CO1: The student would understand the basic	life problems.	and design tool to solve real	CO4. Acquire knowledge	development.	industrial and societal	professional ethics for	economical, and	CO3. Use safe,	lifelong learning.	positive attitude towards	communication skills, and	managerial skills,	CO2. Learn team work,	operations.	analyse the industrial	knowledge to explore and	CO1:Use the theoretical	problems.	chemical engineering	and solve the numerically	CO4:Ability to understand	forms	problems in mathematical	chemical engineering	CO3:Able to convert any	applications.	would be aware about its	chemical engineering and
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50	Computational Fluid Dynamics						Phase –II)	Major Project (								*								
CO2:To understand basic properties of computational	CO1:To understand mathematical characteristics of partial differential equations.	CO4: Ability to address the environmental and societal issues.	communicate effectively.	CO2: Ability to follow professional ethics while preparing project reports.	proposals for growth of profession and society.	knowledge to come up with innovative research	chemical engineering	CO1: Able to use the	the alidience	style and content to match	visual communication and	modes of oral, written and	CO4: To learn the effective	their impact on society.	related to environment and	CO3: To understand the	provide effective solutions.	challenges/limitations to	analysis of these	chemical industries and	challenges/limitations in the	identify current	CO2: To understand and	industries.
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			Multiphase Flow	Introduction to		ž.															Technology	Recovery	Petroleum										
CO3: Reinforcement of	the multiphase flow problem with multiphase flow flow dynamics.	motion equations.	characteristics of	CO1: Understanding of the	industry.	related to petroleum	engineering problems	formulates, and solves	CO3: An ability to identify,	sustainability.	manufacturability, and	health and safety,	social, political, ethical,	economic, environmental,	constraints such as	needs within realistic	process to meet desired	system, component, or	CO2: An ability to design a	interpret data.	as well as to analyze and	and conduct experiments,	CO1: An ability to design	equations	and Navier-Stokes	computationally solve Euler	CO4:To learn how to	differential equations.	various types of partial	solution techniques for	CO3:To learn computational	consistency.	methods-accuracy, stability,
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		CHPE-461															CHPE-459			w 2:
a.		BioChemical Engineering										=		Resources	Energy	Renewable	New and			
CO3Acquire the knowledge of enzyme catalyzed reaction and inhibition mechanisms	CO2Understand the integrated approach of chemical engineering with basic life sciences in developing processes and products.	CO1Understanding of biological basics and bioprocessing	about renewable energy projects	understanding of various possible mechanisms	CO4: Equip the students	instruments used.	concents of equipment and	develop an understanding	CO3: The students also	their application	fundamental concepts and	energy resources the	sufficient knowledge about	CO2: The students acquire	of energy technologies	Non-Conventional sources	among students about	CO1:Create awareness	CO4: Enhancement of team working skills.	knowledge through practice with realistic problems
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	Process Instrumentation and Analystical Methods			Membrane Separation Processes	
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CO2:Students will attain the knowledge of principles for	CO1:The students will get acquaintance about various types of instruments for measurement of temperature, pressure, conductivity, pH, composition of the given mixture etc	CO4: Students are able to identify established membrane separation processes and learn concepts of upcoming membrane separation processes	CO3: The students are capable of applying various transport models for the calculation of membrane fluxes and the other separation properties for various membrane systems	principles for different membrane separation processes  CO2: Identify and design the suitable membrane separation technique for intended problem.	CO4Acquire knowledge about different types of bioreactor, its industrial applications and scale up criteria.
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-				Engineering	Fluidzation								i.						4	60	Engineering	Science and	Interfacial	Colloids and	Introduction to				182							
	CO3: Able to understand the Fluidization of different	fluidized bed systems	Heat and Mass transfer in	and solve Fiuldization.	CO1: Ability to understand	to materials.	applications from processes	translate into a variety of	these concepts and tools	CO4: Appreciation of how	behavior.	exploitation of nano-	regimes and behavior and	surface and bulk dominated	differences between the	CO3:Understanding of	scales.	thermodynamics on small	mechanics and	multi-phase nano-systems;	CO2:Ability to understand	science and engineering.	tools of colloid and interface	nomenclature, concepts and	CO1:Understanding of basic	instruments.	of the industrial	to design and data analysis	CO4:The students will able	applied in analytical tasks.	instrumental methods	in procedures and	students to extend the skills	CO3:The course helps the	chromatographic, etc	spectroscopic and
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concepts about biomass	CO4: Knowing basic concepts of Biorefineries and Green Chemical methods and application to present industrial processes.	CO3: Ability to identify key pathways for sustainable processing of feedstocks	bioproducts from biomass and waste.	Chemical approaches and alternative processes to	CO2: Understanding Green	in a future scenario without fossil fuels.	possibilities of biorefineries	problems	construction and corrosion	knowledge on materials of	CO4: Ability to get	fertilizer industries	problems occurring in	CO3: Able to understand the	various fertilizers	methods of production of	CO2: Able to know different	the importance of fertilizers.	CO1:Ability to understand	Avg	different applications.	fluidized bed system for	CO4: Able to design a	phases.
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	<del>- was set</del>					Assesment	Impact	Environmental												*			and Technology	Nano Science					nuniyi s	,	10						Process
	the current environmental	methods and legislation	the current assessment	CO2Ability to understand	tools used	and the techniques and	the current EIA methods	CO1Ability to understand	nanomaterials	technologies based on	latest devices and	CO4: To get familiar with	nanomaterials	relationship of various	structure and property	CO3: To understandthe	nanomaterials	manufacture of	processes, for the	CO2: To study the synthesis	properties	nanostructures and their	basic concepts of	CO1: To understand the	various processes	production by combining	biofuels and bioenergy	sustainable biorefinery for	CO4: Ability to design a	conversion processes.	various biomasses to energy	CO3: Able to understand the	processing techniques.	pretreatment and	evaluate various biomass	CO2: Understand and	derived energy
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	Sec. Systems			(Biotechnology)	Heat and Mass Transfer				(Jancon qui squ'ill)					Rheology	Industrial								
CO4: The students are able to comprehend the concepts of co current & counter current processes, cascades and concept of Ideal stage and stage	CO3: Ability to understand the principles of mass transport.	dimensional analysis and comprehend the laws governing radiation mode.	CO2: Develop correlations	vection and olems.	and solve conduction,	flows.	industrial non-Newtonian	CO4:Ability to analyze the	in stress and strain controlled experiments	fluids and their behaviour	CO3:Exposure to complex	nonlinear viscoelasticity.	theories of linear and	importance of rheology	CO1: Understanding the	number of case studies	tool with application to a	and prediction as a design	impact modeling	process of environmental	knowledge acquired to the	CO4Ability to apply	monitoring systems
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												(Biotechnology)	Mechanics	Fluid and Particle											2				(Biotechnology)	Control	and Process	Instrumentation		
CO4: The students would understand the seperation	handling of solid-solid and solid-fluid mixtures.	mmeasurement and	understand the physical	CO3: The students would	momentum balance.	application to the	fluid flow problems with the	CO2: Ability to analyze the	compressible fluids.	incompressible and	drop, flow rates etc. for	able to measure pressure	subject, students would be	CO1: After studying the	and allied industries.	control system for chemical	to design the	control systems and be able	identified the stability of	CO4: The students can	controllers	response of various	to understand the transient	CO3:The students will able	higher order systems	first and	various forcing function on	to understand the effect of	CO2:The students will able	block diagram	chemical process in terms of	CO1: To understand the	contact equipment's.	efficiencies, continuous
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